

Laser Ion Source Development for ISOL Systems at RIA

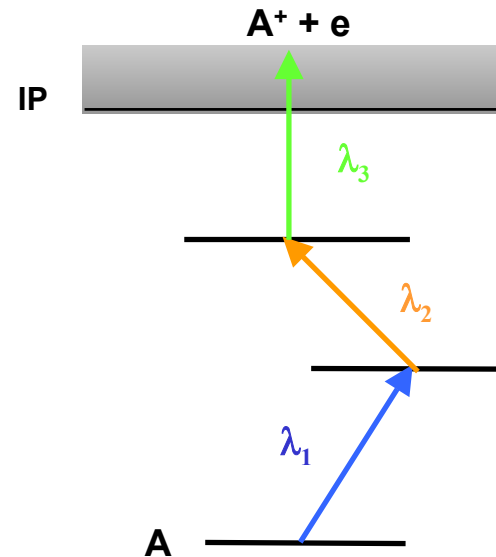
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Laser Ion Source Based On Resonant Photoionization

Ions of a selected element are produced via stepwise resonant photo excitations followed by ionization in the last transition.



- highly selective
- efficient
- versatile -- applicable to about 80% of all the elements in the periodic table (ionization potential $< 10\text{eV}$)

Develop New Ionization Schemes

- Element available at ISOLDE-RILIS
- Dye - Cu - laser RIS successfully tested
- Dye - Cu - laser RIS possible
- Ti:Sa laser RIS possible

<div><div><div></div><div>Dye - Cu - laser RIS successfully tested</div></div><div><div></div><div>Dye - Cu - laser RIS possible</div></div><div><div></div><div>Ti:Sa laser RIS possible</div></div></div>																		<div><div><div></div><div>1H</div></div><div><div></div><div>2He</div></div></div>	
<div><div><div></div><div>3Li</div></div><div><div></div><div>4Be</div></div></div>												<div><div><div></div><div>5B</div></div><div><div></div><div>6C</div></div></div>	<div><div><div></div><div>7N</div></div><div><div></div><div>8O</div></div></div>	<div><div><div></div><div>9F</div></div><div><div></div><div>10Ne</div></div></div>					
<div><div><div></div><div>11Na</div></div><div><div></div><div>12Mg</div></div></div>												<div><div><div></div><div>13Al</div></div><div><div></div><div>14Si</div></div></div>	<div><div><div></div><div>15P</div></div><div><div></div><div>16S</div></div></div>	<div><div><div></div><div>17Cl</div></div><div><div></div><div>18Ar</div></div></div>					
<div><div><div></div><div>19K</div></div><div><div></div><div>20Ca</div></div></div>	<div><div><div></div><div>21Sc</div></div><div><div></div><div>22Ti</div></div></div>	<div><div><div></div><div>23V</div></div><div><div></div><div>24Cr</div></div></div>	<div><div><div></div><div>25Mn</div></div><div><div></div><div>26Fe</div></div></div>	<div><div><div></div><div>27Co</div></div><div><div></div><div>28Ni</div></div></div>	<div><div><div></div><div>29Cu</div></div><div><div></div><div>30Zn</div></div></div>	<div><div><div></div><div>31Ga</div></div><div><div></div><div>32Ge</div></div></div>	<div><div><div></div><div>33As</div></div><div><div></div><div>34Se</div></div></div>	<div><div><div></div><div>35Br</div></div><div><div></div><div>36Kr</div></div></div>											
<div><div><div></div><div>37Rb</div></div><div><div></div><div>38Sr</div></div></div>	<div><div><div></div><div>39Y</div></div><div><div></div><div>40Zr</div></div></div>	<div><div><div></div><div>41Nb</div></div><div><div></div><div>42Mo</div></div></div>	<div><div><div></div><div>43Tc</div></div><div><div></div><div>44Ru</div></div></div>	<div><div><div></div><div>45Rh</div></div><div><div></div><div>46Pd</div></div></div>	<div><div><div></div><div>47Ag</div></div><div><div></div><div>48Cd</div></div></div>	<div><div><div></div><div>49In</div></div><div><div></div><div>50Sn</div></div></div>	<div><div><div></div><div>51Sb</div></div><div><div></div><div>52Te</div></div></div>	<div><div><div></div><div>53I</div></div><div><div></div><div>54Xe</div></div></div>											
<div><div><div></div><div>55Cs</div></div><div><div></div><div>56Ba</div></div></div>	<div><div><div></div><div>57La</div></div><div><div></div><div>72Hf</div></div></div>	<div><div><div></div><div>73Ta</div></div><div><div></div><div>74W</div></div></div>	<div><div><div></div><div>75Re</div></div><div><div></div><div>76Os</div></div></div>	<div><div><div></div><div>77Ir</div></div><div><div></div><div>78Pt</div></div></div>	<div><div><div></div><div>79Au</div></div><div><div></div><div>80Hg</div></div></div>	<div><div><div></div><div>81Tl</div></div><div><div></div><div>82Pb</div></div></div>	<div><div><div></div><div>83Bi</div></div><div><div></div><div>84Po</div></div></div>	<div><div><div></div><div>85At</div></div><div><div></div><div>86Rn</div></div></div>											
<div><div><div></div><div>87Fr</div></div><div><div></div><div>88Ra</div></div></div>	<div><div><div></div><div>89Ac</div></div><div><div></div><div>104Rf</div></div></div>	<div><div><div></div><div>105Ha</div></div><div><div></div><div>106</div></div></div>	<div><div><div></div><div>107</div></div><div><div></div><div>108</div></div></div>	<div><div><div></div><div>109</div></div><div><div></div><div>110</div></div></div>	<div><div><div></div><div>111</div></div><div><div></div><div>112</div></div></div>	<div><div><div></div><div>113</div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div><div></div><div></div></div><div><div></div><div></div></div></div>				

58Ce	95Pr	60Nd	61Pm	62Sm	63Eu	64Gd	65Tb	66Dy	67Ho	68Er	69Tm	70Yb	71Lu
90Th	91Pa	92U	93Np	94Pu	95Am	96Cm	97Bk	98Fm	99Es	100Fm	101Md	102No	103Lr

**Laser ion sources are being used or developed
for on-line production of RIBs at several ISOL facilities**

ISOLDE (CERN)

ISAC/TRIUMF

GSI

Institute of Spectroscopy (Troitsk)

IRIS (Gatchina)

Mainz University

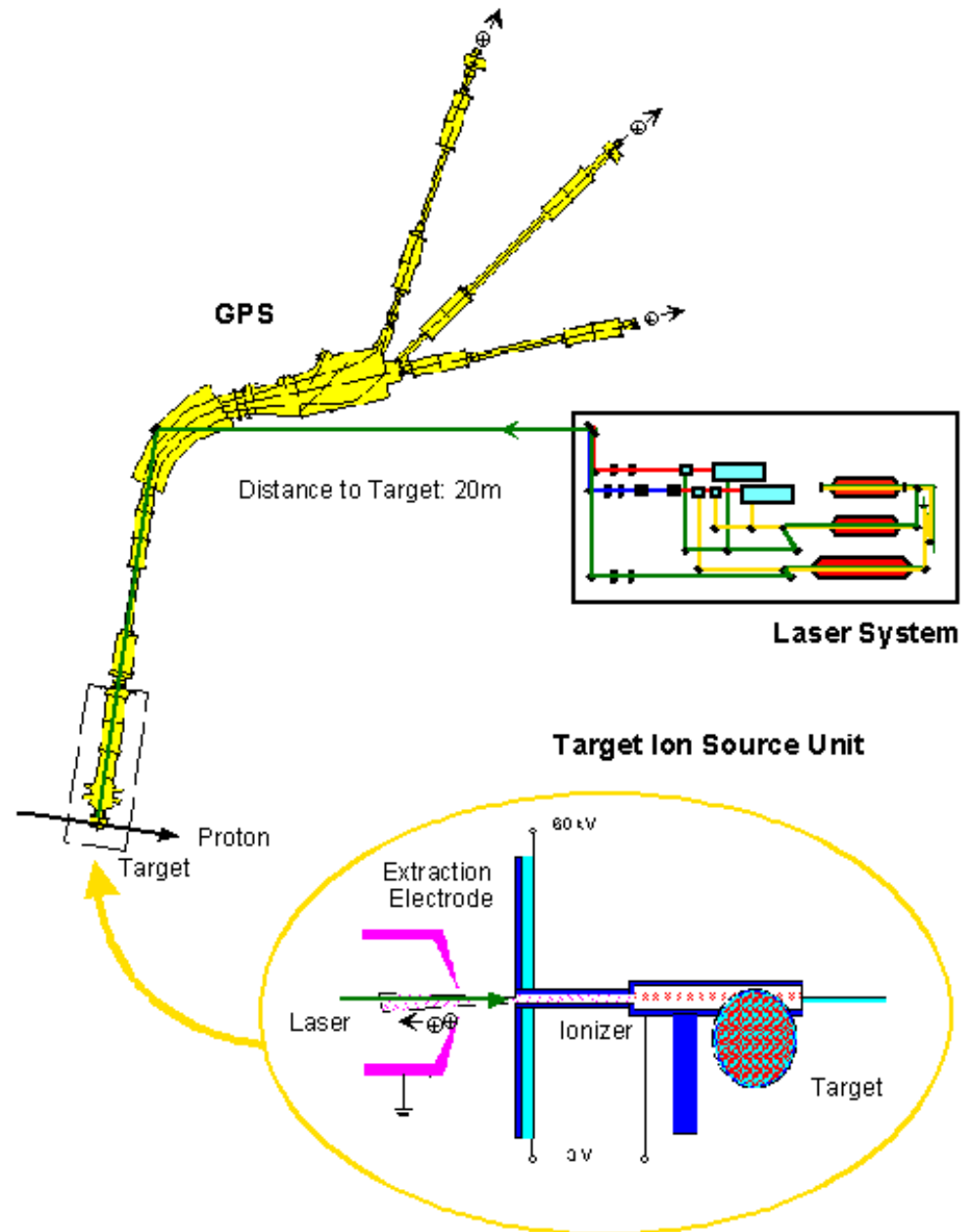
LISOL, Louvain-la-Neuve

TIARA (Takasaki)

Heavy Ion Research Facility (Lanzhou, China)

ISODE RILIS

- Copper vapor laser pumped dye lasers
- Hot cavity interface
- > 20 RIBS provided with overall efficiencies of the order of 10%



Areas in which improvements and developments can be made

- **Efficiency**
 - Higher laser power in the ionizer cavity
 - Improved ion extraction
 - Development of ionization schemes utilizing autoionizing states instead of ionizing directly to the continuum
 - Fractionated ground-state population
- **Selectivity**
 - Ionization via laser excitation competes with surface ionization in a hot cavity
 - Investigate schemes to ionize refractory elements that may be transported as molecules
- **Reliability and Maintenance**
 - Use solid state pump lasers instead of copper vapor lasers
 - Use Ti:sapphire lasers to replace dye lasers (a gap in wavelength of 500-700 nm will result)

Requirements for a new Laser Ion Source

- **High power**
- **High repetition rate (> 10 kHz)**
- **High reliability**
- **Easy tunability**
- **Reduction of consumables and maintenance**
- **Cover a wide range of wavelengths**
- **Capability for 2-step and 3-step ionization schemes**

Closing Comments

- **This research can be accomplished at the Holifield Radioactive Ion Beam Facility (HRIBF) in an operating ISOL environment with both stable isotopes and short-lived radioactive isotopes**
- **At ORNL, laser expertise exists in the atomic physics group and among some members of the group presently working in ion source development**
- **Useful discussions have been initiated with Ulli Koester (ISOLDE) concerning what directions/problems a new laser ion source development project should focus on**
- **The plan is not to develop new laser technology but to use state-of-the-art laser technology to improve the laser and ion source coupling, develop new ionization schemes, and develop a system that is reliable, efficient, easily tuned, and easy to maintain**